

Oh Were They Ever Happy Book

R Sandford

Question #18488 - Socratic The degree of dissociation α (0.0158) $K_b = 2.51 \times 10^{-6}$ color (white) (x) "mol/l") Triethylamine is a weak base and ionises: $(CH_3)_3N + H_2O \rightleftharpoons (CH_3)_3NH^+ + OH^-$... **Question #71ce2 - Socratic** Nov 9, 2014 · $H^+ + OH^- \rightarrow H_2O$ when the acid was added to the resulting solution. The H^+ and OH^- react in a 1:1 ratio. This tells us that the number of moles of H^+ used will be equal ... *What is the chemical formula for lithium hydroxide? - Socratic* Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M^+ ion. Hydroxide anion, OH^- , has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry ... *Question #22e51 - Socratic* Products formed are; $Cu(OH)_2 + HCl \rightarrow CuCl_2 + H_2O$ Since $CuCl_2$ is soluble in water.. But; $Cu(OH)_2 + HCl \rightarrow CuCl_2 + H_2O$ " [Not Balanced]" Balancing the equation.. What is the product of the following reaction? 1) CH_3OH - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... $NH_4^+ + HO^- \rightleftharpoons NH_3(aq) + H_2O(l)$ For methanol, the acid base reaction would proceed ... **What are the spectator ions in the reaction between sodium** Oct 16, 2016 · The nitrate and the sodium ions. $Na_2CO_3(aq) + 2AgNO_3(aq) \rightleftharpoons Ag_2CO_3(s) + 2NaNO_3(aq)$ The net ionic equation is: $2Ag^+(+) + CO_3^{2-} \rightleftharpoons ...$ **Could a buffered solution be made by mixing aqueous ... - Socratic** Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered ... **Question #627ef - Socratic** functional groups like -OH, -CHO, -CO- are present in carbohydrates. commonly two functional are present in a carbohydrate, -OH and -CHO, or -OH and -CO-. Aldose shows different reactions ... **Question #44ec4 - Socratic** That's a silly way to say carbon dioxide! $CO_2(g) + Ca(OH)_2(aq) \rightleftharpoons Ca^{2+}(aq) + H_2CO_3(aq)$ May be wrong, but it seems reasonable. The gas would interact ... **In each of the following pairs, which one has the lower viscosity?** Jul 31, 2015 · C_2H_5Cl has weak dipole-dipole forces from the C-Cl bonds, but CH_3OH has strong hydrogen bonds. Confirmation: The viscosity of C_2H_5Cl is 0.279 mPa·s, while the viscosity of ...

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