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fraction $n_a = 2 \text{ hpo} \times 4.0825 \text{ g} \times 141.9579 \text{ g/mol} \times 0.00581158 \text{ mol} \times 2 \times 0.4500 \text{ g} \times 18.015 \text{ g/mol} \times 24.97918401 \text{ mol}$

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calcium hydroxide Ca(OH)_2 in 40.00 ml of solution 10.5 kg
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