

A La Sombra Del Angel

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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$3\text{Na}^+ + \text{AlPO}_4 + 3\text{KOH} \rightarrow 3\text{K}^+ + \text{AlPO}_4 + 3\text{OH}^-$
 $\text{FeCl}_3 + 3\text{Mg} \rightarrow 3\text{MgCl}_2 + 2\text{Fe}$

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