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Jicheng Xie

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carbon carbon bond in this

condensation an enol or enolate ion

reacts with a carbonyl compound to

form a  $\beta$  hydroxyketone or  $\beta$

hydroxyaldehyde which is then followed by dehydration

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dibenzalacetone by aldol condensation

19 py the aldol condensation is a

reaction between two aldehydes or

ketones catalyzed by a base or acid

generating a molecule having both

alcohol and aldehyde functional groups

the aldol product is either a  $\beta$

hydroxyaldehyde or a  $\beta$  hydroxyketone

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occurs via an aldol condensation

reaction in fig 2 figure 2 shows the

reaction of an aldehyde with a ketone

employing sodium hydroxide as the

base which is an example of an aldol

condensation mixed reaction

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dibenzylideneacetone involves the aldol

condensation of acetone and

benzaldehyde the reaction requires two

equivalents of benzaldehyde relative to

acetone in order to form the final

product the reaction is driven largely

by the insolubility of

dibenzylideneacetone in the reaction

solvent of ethanol and water scheme 2

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condensation is a condensation reaction

in organic chemistry in which an enol

or an enolate ion reacts with a carbonyl

compound to form a  $\beta$  hydroxyaldehyde

or  $\beta$  hydroxyketone followed by

dehydration to give a conjugated enone

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condensation purpose synthesize dibenzalacetone which is aldol compound by claisen schmidt reaction to prove that reaction of an acetone with aldehyde catalyzed by a strong base yields an aldehyde alcohol mixed compound known as aldol [aldol condensation synthesis of dibenzalacetone yumpu](#) - Mar 03 2022 web dec 7 2015 experiment we will prepare the dibenzalacetone 1 5 diphenyl 1 4 pentadien 3 one the equilibrium is shifted toward the product because the compound precipitates from the reaction mixture as it is formed hhhoh1 5 diphenyl 1 4 pentadien 3 one strong dibenzalacetone strong is a common ingredient in sunscreen

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web this experiment was being performed so that dibenzalacetone could be synthesized from benzaldehyde and acetone this experiment was performed to show how a ketone and an aldehyde could be added together through the aldol condensation the aldol condensation is extremely important because it can

form a  $\beta$  hydroxy aldehyde or

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web introduction aldol condensations are essential to organic synthesis because they are able to produce carbon carbon bonds an enolate anion is formed starting aldol condensation the enolate anion comes from an aldehyde or a ketone enolizable aldehydes and ketones undergo a reaction when either an acid or base catalyst is present

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web an aldol reaction was performed by combining acetone and benzaldehyde to make dibenzalacetone the percent yield of dibenzalacetone was found to be 73 which is a good percent yield the melting point of dibenzalacetone was around the range of 104 110 c this helped conclude that dibenzalacetone had a trans trans e e isomer

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