

# The American Scholar Pdf

## SB Merriam

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6 3s 2 3p 4 88 chlorine 17 1s 2 2s 2 2p

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number of 8 give its electron

configuration answer an atomic number

of 8 means a total of eight electrons

two of these electrons can fit into the

first shell while the remaining six can

fit into the second shell therefore the

configuration is written as 2 6

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s p d and f have different shapes and

one orbital can hold a maximum of two

electrons the p d and f orbitals have

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configuration of each element is unique

to its position on the periodic table

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fit into the 1s orbital 1s 2 then next two

into the 2s orbital 2s 2 the 2 porbital

has three slots which can take 2

electrons each a total of 6 which gives

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of atoms is key but also keep in mind

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electron to leave the atom would be one

in the highest energy level which is the

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highest energy level are the least attracted to the nucleus because they are the most distant 25

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1s22s22p63s23p64s24d104p65s1 no it should be 3d10

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