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**Principios de química** Richard E. Dickerson,1992-11 Esta tercera edición de Principios de Química, se ha proyectado para ser utilizada en un curso universitario de Química general, que debe servir tanto como una visión de la Química para los especialistas, como de una buena base para los estudios posteriores de las disciplinas de Química. Por consiguiente, hay varios capítulos que pretenden introducir las diferentes áreas de la Química, incluida la inorgánica, nuclear, orgánica y Bioquímica, y se intenta, a lo largo de todo el libro, colocar a la Química en su marco histórico y cultural. Al mismo tiempo, se presentan los aspectos cuantitativos de la Química de forma consecuente con su importancia, de manera que resulte fácil apoyarse en ellos en los cursos posteriores.

Química en ejercicios Programa UBA XXI // María José Billorou et al. Universidad de Buenos Aires,2014-03-01 Ejercicios de química para estudiantes de UBA XXI

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Quimica: la Ciencia Central Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish.Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry.For anyone interested in learning more about chemistry.

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*Química 2 (SEP) ,*

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**Técnica y síntesis en química inorgánica** Robert J. Angelici, 1979 La mayor parte de las técnicas que consideramos importantes han sido incluidas en este Manual:

*Fundamentos de la química general e inorgánica* Hans Rudolf Christen, 1977 Este texto tiene como objetivo proporcionar a los estudiantes una introducción en los fundamentos teóricos de nuestra ciencia que les permita alcanzar simultáneamente una primera versión general sobre el campo de la Química inorgánica. De acuerdo con esta idea, la parte general -teoría atómica y de enlaces, termodinámica, cinética, reacciones ácido-base y redox, Química de los complejos- ocupa un espacio bastante amplio, mientras que la Química descriptiva -campo del estudio de las sustancias propiamente dicho- queda más bien en un segundo término sin formar el objetivo fundamental de la obra.

*Física Y Química. Vol. Iii: Química I. Profesores de Educacion Secundaria. Temario Para la Preparacion de Oposiciones. Ebook , 2000-09-13*

**Introducción a la química** Leo J. Malone, 2006 Química : materia, cambios y energía - Mediciones en química - El átomo, la estructura de la materia y las reacciones nucleares - El arreglo de los electrones en el átomo - La naturaleza periódica de los elementos - La naturaleza del enlace químico - La nomenclatura de los compuestos - Relaciones cuantitativas : la mol - Reacciones químicas : ecuaciones y cantidades - El estado gaseoso de la materia - Agua : los estados líquidos y sólidos - Soluciones acuosas - Ácidos, bases y sales - Reacciones de oxidación-reducción - Cinética y equilibrio químico -

Introducción a la química orgánica.

**Contactos** ,1997

*Química, una versión moderna* Guido Guayasamin Guerrero,1979

Boletín PPDQ ,2001

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