

# I Am The Lorax

J Rink

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the main difference between actinides  
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below the lanthanide series which is  
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rare earth metals these elements all  
have a high diversity in oxidation  
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radioactive

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actinides are radioactive in nature they

consist of 30 elements in total they contain radioactive elements such as uranium these elements are kept in the lanthanide and actinide series of periodic tables here we will learn what are the lanthanides and actinides

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web on the basis of the entrance of the last electron either into 4 f orbital or 5 f orbital these elements are classified into two categories called lanthanides and actinides respectively lanthanides the name lanthanides comes from lanthanum la z 57 because these elements are closely related to lanthanum

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am cm from the lanthanides at an  
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metals and have properties of both the  
d block and the f block elements but  
they are also radioactive lanthanides  
have different chemistry from transition  
metals because their 4f orbitals are  
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filled with electrons and the actinides  
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generally speaking the lanthanides

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notation in the main body cell rather  
than making room for all these  
elements in the 3b group in the main



body of the periodic table because of the f shell electrons these two element groups are also known as the f block elements

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