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 3h

web identify the missing particle in the following nuclear reaction

$${}_{10}^{235}\text{U} + {}_0^1\text{n} \rightarrow {}_{40}^{96}\text{Zr} + {}_{38}^{96}\text{Sr} + 2{}_0^1\text{n} + 4{}_2^4\text{He} + {}_{52}^{137}\text{Te}$$

web the coefficient in a balanced equation is an idea the concept of telling the chemist that if the atoms molecules and compounds are balanced there are balanced amounts of the atoms molecules and compounds on the other side in the product

web answer 4 correct in the reactants and products must be the same part b what is the value of in the following nuclear reaction 6 5i 9 express your answer as an integer hint 1 how to approach the problem use rule 2 the

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equations the sums of the atomic and mass numbers must be the same on both sides of the equation in some cases one of the symbols in the list below will be used to complete the equation

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