Oh The Places Youll Go

J Spring

What is the product of the following reaction? 1)CH 3 OH - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... NH $4^+ + HO^-$ (-) rarr NH 3(ag) + H 2O(1) For methanol, the acid base reaction would proceed ... **0.10 moles of sodium cyanate dissolved in 250cm^3 distilled** Dec 28, 2015 · The cyanate anions will react with water to form isocyanic acid, HNCO, and hydroxide anions, OH-. This tells you that you can expect the pH of the resulting solution to be higher than 7. Calculate the molarity of the sodium cyanate solution by using the given number of moles and volume. Keep in mind that you need to use liters of solution and ... Could a buffered solution be made by mixing aqueous ... - Socratic Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered solution? What is the mass percent of hydrogen for C 3H 5OH? | Socratic Jan 2, 2017 · In order to compute the mass percent of hydrogen in C3H 5OH we're going to need two things: 1: The formula mass of the entire compound 2: The atomic mass of hydrogen The formula mass of C3H 5OH is 58.08 g mol because: 3(12.01 g mol) + 6(1.008 g mol) + (16.00 g mol) = 58.08 g mol The atomic mass of the hydrogen atom is 1.008 g mol Now, we have to use the following equation: The numerator ... If the pH=9 for a solution of KOH(ag), what is the ... - Socratic Aug 18, 2016 · If the pH = 9 for a solution of K OH (ag), what is the concentration of HO-? What is the chemical formula for lithium hydroxide? - Socratic Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M + ion. Hydroxide anion, -OH, has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry between ions: Li(s) + H 2O(l) \rightarrow LiOH (aq) + 1 2 H 2(q) $\uparrow \sqcap \sqcap \sqcap$ What is the formula of "magnesium hydroxide"? | Socratic Feb 27, 2017 · Mg (OH) 2 Now we know that hydroxides are salts of HO^-, and some metal ion. Now if the parent metal has an electronic configuration of 2:8:2, then there are 12 electrons, and the atomic number of the metal is equal to 12. We look on the Periodic Table, and we find that Z=12, for "magnesium metal". As a Group 2 metal, magnesium forms a Mg^ (2+) ion, and hence its hydroxide is Mg (OH) 2. What are the spectator ions in the reaction between sodium Oct 16, 2016 · The nitrate and the natrium ions. Na 2CO 3(ag) + 2AgNO 3(ag) rarr Ag 2CO 3(s)darr + 2NaNO 3(ag) The net ionic equation is: 2Ag^(+) + CO 3^(2-) rarr Ag 2CO 3(s)darr ... **Question** #44ec4 - Socratic That's a silly way to say carbon dioxide! CO 2(g)+Ca(OH) 2(ag) rightleftharpoons Ca^(2+)(ag)+H 2CO 3(ag) May be wrong, but it seems reasonable. The gas would interact with the hydroxide and water

molecules to form carbonic acid. There are also concentrations of OH^(-) and H^(+) leftover from this reaction and from the auto-ionization of water, but that's probably going too far into detail. **Question #627ef - Socratic** functional groups like - OH, -CHO,-CO- are present in carbohydrates. commonly two functional are present in a carbohydrate,-OH and -CHO, or -OH and-CO-. Aldose shows different reactions and test like Tollen's silver mirror test, felling's test. different types of reactions are also shown by carbohydrates like Osazone synthesis, length increasing reaction, ...

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