

# Propiedades De Las Bases Y Acidos

Juan A. Lloréns Molina

*Acidos Y Bases ,*

*Química. La ciencia básica* DOMINGUEZ REBOIRAS, MIGUEL ANGEL,2006-01-01 Esta obra se ha escrito pensando en las necesidades de los estudiantes de los primeros cursos de Facultades de Ciencias, Farmacia, Ciencias Ambientales y de la Salud, y Escuelas Tecnicas.

**Química. Fundamentos experimentales** Robert W. Parry,1973 Este libro es una versión autorizada del célebre libro original CHEM Study, cuya versión española se ha difundido por los Centros de Enseanza donde se cursan estudios de nivel medio y preuniversitario.Tanto en este libro, como en el CHEM, la experimentación es el vehículo para presentar la Química, tal como es hoy en día.Los principios unificadores se desarrollan a partir de la observación experimental. Así, la Química aparece como una Ciencia y no como una masa de información. El objetivo principal de este libro es lograr fundamentalmente el hito de la Ciencia moderna: desarrollo del principio a partir de la observación.

**Química** Paul R. O'Connor,1977 Este libro, es junto con el Manual de laboratorio, fruto directo del programa CHEM publicado también por esta Editorial. Como una de las versiones autorizadas del citado proyecto, esta obra refleja todo el esfuerzo y dedicación del equipo original del proyecto CHEM. El título Química. Experimentos y teorías responde perfectamente bien a la idea básica de este libro; en él se exponen cuidadosamente y además se utilizan a lo largo del mismo, todos los pasos por los cuales transcurre el llamado método científico. Las observaciones experimentales y las medidas dan lugar al desarrollo de los principios teóricos que las unifican y que, más tarde, se utilizan para relacionar e interpretar diversos fenómenos.

*Química* Ronald J. Gillespie,1988-08

**Química, un proyecto de la A.C.S.** American Chemical Society,2005 Este texto de Química está desarrollado y escrito por un grupo de excelentes educadores químicos escogidos por la American Chemical Society. Poniendo un constante énfasis en el aprendizaje activo, la obra propone a los estudiantes que no lean simplemente sobre el mundo de la química, sino que lo exploren de manera personal. Química es un libro esencial para aquellos profesores que quieran incorporar a su docencia el aprendizaje basado en la investigación.

**Química 1. Fundamentos ,**

## **Bioquímica Estructural** José María Teijón Rivera, Manuel Ruiz Amil, 1992-06

*Química*, 2005 CONTENIDO: Introducción a la química - Sistema de medidas - Materia y energía - La estructura del átomo - Clasificación periódica de los elementos - Estructura de los compuestos - Nomenclatura química de los compuestos inorgánicos - Cálculos que comprenden elementos y compuestos - Ecuaciones químicas - Cálculo en las ecuaciones químicas. Estequiometría - Gases - Líquidos y sólidos - Agua - Disoluciones y coloides - Ácidos, bases y ecuaciones iónicas - Ecuaciones de oxidación-reducción y electroquímica - Velocidades de reacción y equilibrio químico - Química orgánica - Química nuclear.

Química. Fundamentos experimentales. Guía del profesor Robert W. Parry, 1974 Esta Guía está destinada a proporcionar apoyo al profesor y a establecer claramente los principios y las razones que guiaron a los autores en la programación y desarrollo del libro Química, fundamentos experimentales.

Trastornos ácido-base Rodrigo Sepúlveda, 2024-03-06 Trastornos ácido-base. Fisiología, patología y clínica es un texto fundamental para todo estudiante y profesional de la salud que deba enfrentarse a problemas ácido-base. En estas páginas el lector podrá aprender y consultar todos los aspectos relacionados a los trastornos ácido-base: química básica, fisiología y fisiopatología, manifestaciones clínicas, exámenes de laboratorio, interpretación integral de los trastornos, herramientas diagnósticas, causas y sus tratamientos, manejo del paciente crítico y ambulatorio, diversos métodos de diagnóstico, y mucho más. Esta publicación cuenta con múltiples recursos gráficos, tablas y esquemas que facilitan su comprensión. En su estructura consta de capítulos conectados entre sí, que van desde lo básico hasta lo más complejo; además se incluyen varios ejercicios con casos clínicos reales de interpretación ácido-base. Debido a que los trastornos ácido-base repercuten en todas las áreas de las ciencias biológicas, la información relacionada a estos se encuentra dispersa, desordenada y sin conexión. Este libro unifica y ordena este conocimiento, entregándolo al lector de manera conectada y gradual, para que pueda aprender y desenvolverse en forma óptima ante cualquier situación clínica.

## Manual de física y elementos de química Manuel Rico y Sinobas, Mariano Santisteban, 1856

*Conocer Los Materiales* Juan A. Lloréns Molina, 1996 Este libro tiene como objetivo proponer al profesorado de la E.S.O. ejemplos de actividades de las áreas de Ciencias de la Naturaleza y Tecnología.

**Bioquímica** Antonio Peña Díaz, 1980 CONTENIDOS : Átomos, iones y moléculas, componentes de los seres vivos - El agua, componente principal de los seres vivos - Las proteínas, maquinaria de la vida - Los lípidos - Los carbohidratos, almacenes de energía solar - Los ácidos nucléicos, moléculas de la herencia - Las enzimas : aceleradores de las reacciones bioquímicas - Bioenergética - Los caminos metabólicos de los carbohidratos - Los caminos metabólicos de los lípidos - Metabolismo de los aminoácidos: producción de energía y síntesis de aminas y otros compuestos nitrogenados - La información genética - Rearreglos genéticos - La célula y el organismo: interacciones celulares internas y externas - Las bases bioquímicas de la nutrición.

**Fundamentos de bioquímica estructural** José María Teijón,2006 Con la idea de facilitar la comprensión de los procesos y mecanismos vitales de los organismos a los estudiantes de las licenciaturas y diplomaturas de Ciencias de la Salud, un equipo de catedráticos y profesores de diferentes universidades, han plasmado su experiencia docente en estos dos tomos de Fundamentos de Bioquímica. El primer tomo se dedica a los aspectos estructurales, y en él se describen las sustancias, sus propiedades y las funciones que realizan en los organismos. En el segundo tratan los aspectos metabólicos y se estudian las transformaciones de las sustancias y los procesos energéticos que las acompañan y que sirven para el funcionamiento normal de los organismos. Al inicio de cada tema se incluye una introducción que fija los objetos a cumplir y, al final de cada capítulo, un resumen repasa los conceptos fundamentales tratados en el capítulo. También se incluye un apartado dedicado a las diversas aplicaciones clínicas en las que se describen algunos casos prácticos relativos al contenido de cada tema. Un libro imprescindible para el docente y estudiante de Ciencias de la Salud, fruto de la experiencia en la docencia en Bioquímica del prestigioso equipo de autores coordinado por los catedráticos Amando Garrido y José María Teijón, escrito con el deseo de crear interés y entusiasmo por esta materia.

**Principios de Química. La búsqueda del conocimiento.** P. W. Atkins,Loretta Jones,2006

*Química básica* Alejandrina Gallego Picó,2018-10-08 El texto, básico pero con rigor, ofrece una exposición de los principios de la Química, accesible incluso para estudiantes sin apenas conocimientos de esta materia. A su parte teórica se une una completa y amplia colección de ejercicios, algunos explicados detalladamente y resueltos; y, otros, propuestos e indicada su resolución, Los autores de este libro son profesores de la UNED y han elaborado diversos materiales didácticos adaptados a la enseñanza a distancia de materias regladas y no regladas.

*Química general* Ramírez Regalado, Víctor Manuel,2020-05-13 Química general. 2<sup>a</sup>. Ed. Esta obra se diseñó con base en los programas de estudio actualizados de la Escuela Nacional Preparatoria de la UNAM. Esta materia abarca un curso anual y básico para el estudio de la química. La obra incluye los contenidos conceptuales, procedimentales y actitudinales que marca la ENP-UNAM. Toda la información está totalmente actualizada y el libro cuenta con una gran variedad de recursos didácticos.

**Química. Ciencia experimental** Chemical Education Material Study,1987 El programa CHEM centra su problemática en la observación y en la experimentación y destaca la importancia de aprender a observar, dedicando más de treinta páginas a sacar todo el fruto posible de la observación que realizan unos alumnos del proceso de combustión de una vela.Se utilizan las interpretaciones teóricas según conviene, pero se insiste constantemente en que lo más importante es tener en cuenta que los hechos químicos, correspondientes a fenómenos más complicados que los hechos físicos, necesitan que el alumno reciba un adiestramiento especial para que no escapen a su percepción.

*Quimica: la Ciencia Central* Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The

scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish. Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

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