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web determine the final temperature and state of the ammonia once the system reaches equilibrium assuming that no heat is lost to the surroundings use the information below

normal boiling point of H_2O is 100°C ΔH_{vap} is 40.7 kJ mol^{-1}
 specific heat of H_2O is $4.18 \text{ J g}^{-1} \text{ }^\circ\text{C}^{-1}$ specific heat of H_2O is $2.06 \text{ J g}^{-1} \text{ }^\circ\text{C}^{-1}$
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web once all of the material has solidified at the time indicated by point c the cooling will continue at a rate determined by the heat capacities of the two solids a and b the composition and of course the geometry of the experimental set up by measuring cooling curves for samples of varying composition one can map the entire phase [heating and cooling curves example 1 channels for pearson](#) - Feb 27 2023

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