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**Tobias Bleicker**

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web sep 17 2017 the chemical elements in the actinide series are called actinides and the elements in the lanthanide series are called lanthanides the main difference between actinides and lanthanides is that actinides can form

complexes easily whereas lanthanides do not form complexes easily key areas covered 1 what are actinides

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web jan 13 2023 the series is the row below the lanthanide series which is located underneath the main body of the periodic table lanthanide and actinide series are both referred to as rare earth metals these elements all have a high diversity in oxidation numbers all of the actinides are radioactive

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web on the basis of the entrance of the last electron either into 4 f orbital or 5 f orbital these elements are classified into two categories called lanthanides and actinides respectively lanthanides the name lanthanides comes from lanthanum la z 57 because these elements are closely related to lanthanum

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web the lanthanides and actinides including group 3 metals due to their chemical similarities account for around a third of the periodic table and they are key to a wide range of basic science research and societally beneficial technologies since 2019 is designated as international year of the periodic table it is an appropriate time to

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web actinides in the periodic table part of a series on the periodic table periodic table forms periodic table history sets of elements by periodic table structure by metallic classification metals alkali alkaline earth transition post transition lanthanide actinide superactinide metalloids dividing metals and nonmetals nonmetals unclassified *transforming lanthanide and actinide chemistry with nanoparticles* - Mar 26 2023

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6 14 *lanthanides and actinides chemistry libretexts* - Aug 31 2023

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web dec 1 2022 lanthanides and actinides vary in electron filling order edit edit source the lanthanides and the actinides make up the f block of the periodic table the lanthanides are the elements produced as the 4f sublevel is filled with electrons and the actinides are formed while filling the 5f sublevel generally speaking the lanthanides

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