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Jesús Biel Gayé

Curso Sobre el Formalismo y Los Métodos de la Termodinámica Jesús Biel Gayé, 1997 Consultar comentario general de la obra completa.

Principios básicos y cálculos en ingeniería química David M. Himmelblau, 1997

Resolución de problemas de química general R. Nelson Smith, Conway Pierce, 1991 Este libro será útil a aquellos alumnos que empiezan a estudiar Química y tienen dificultades al analizar los problemas y encontrar soluciones lógicas o que tienen dificultades con las representaciones gráficas y su interpretación.

Fundamentos de tecnología química Vollrath Hopp, 2022-05-05 Fundamentos de tecnología química es un libro de texto y de enseñanza al servicio de las jóvenes generaciones que desean adquirir una formación profesional y especialmente para los que buscan una cualificación en las actividades propias de las fábricas químicas.

Bioquímica Jeremy Mark Berg, Lubert Stryer, John L. Tymoczko, 2007 Esta nueva edición del Stryer mantiene todas las características que han hecho de esta obra un auténtico best-seller en la enseñanza de la materia. Las razones de este éxito son diversas: texto claro y riguroso, contenido amplio y actualizado, ilustrado, ilustraciones didácticas y atractivas... En esta sexta edición se han enriquecido y actualizado sus contenidos, destacando las siguientes aportaciones: se pone un mayor énfasis en los aspectos fisiológicos, se ofrece una perspectiva molecular de la evolución y se incluyen nuevos capítulos sobre la hemoglobina y el desarrollo de fármacos, así como nuevas aplicaciones clínicas y mejores visualizaciones de las estructuras moleculares. Es de destacar el material complementario incluido en la espectacular página web del libro (www.whfreeman.com/stryer y próximamente en www.reverte.com/stryer), muy útil para facilitar el proceso de aprendizaje.

Química, 2005 CONTENIDO: Introducción a la química - Sistema de medidas - Materia y energía - La estructura del átomo - Clasificación periódica de los elementos - Estructura de los compuestos - Nomenclatura química de los compuestos inorgánicos - Cálculos que comprenden elementos y compuestos - Ecuaciones químicas - Cálculo en las ecuaciones químicas. Estequiometría - Gases - Líquidos y sólidos - Agua - Disoluciones y coloides - Ácidos, bases y ecuaciones iónicas - Ecuaciones de oxidación-reducción y electroquímica - Velocidades de reacción y equilibrio químico - Química orgánica - Química nuclear.

Química: teoría y problemas José María Teijón, José Antonio García Pérez, 1996 Explica cada uno de los conceptos de la Química General y los aplica en más de 500 problemas resueltos. Incluye prácticas de laboratorio y tests de autoevaluación.

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Bioquímica. Con aplicaciones clínicas Thomas M. Devlin, 2015 Parte I. Estructura de las macromoléculas 1. Estructura celular eucariótica 2. DNA y RNA: Composición y estructura 3. Proteínas I: Composición y estructura Parte II. Transmisión de la información 4. Replicación, recombinación y reparación del DNA 5. RNA: Transcripción y maduración del RNA 6. Síntesis de proteínas: Traducción y modificaciones postraduccion 7. DNA recombinante y biotecnología 8. Regulación de la expresión génica Parte III. Funciones de las proteínas 9. Proteínas II: Relación estructura-función de familias de proteínas 10. Enzimas: Clasificación, cinética y control 11. Los citocromos y las óxido nítrico sintasas 12. Membranas biológicas: Estructura y transporte a través de membranas. Parte IV. Rutas metabólicas y su control 13. Bioenergética y metabolismo oxidativo 14. Metabolismo glucídico I: Principales rutas metabólicas y su control 15. Metabolismo glucídico II: Rutas especiales y gluconjugados 16. Metabolismo lipídico I: Utilización y almacenamiento de energía en forma de lípidos 17. Metabolismo lipídico II: Rutas metabólicas de lípidos especiales 18. Metabolismo de los aminoácidos 19. Metabolismo de los nucleótidos purínicos y pirimidínicos 20. Interrelaciones metabólicas Parte V. Procesos fisiológicos 21. Bioquímica de las hormonas I: Hormonas polipeptídicas 22. Bioquímica de las hormonas II: Hormonas esteroides 23. Biología molecular de la célula 24. Metabolismo del hierro y del hemo 25. Digestión y absorción de los constituyentes básicos de la nutrición 26. Principios de nutrición I: Macronutrientes 27. Principios de nutrición II: Micronutrientes Apéndice - Repaso de Química Orgánica.

Propiedades mecánicas y térmicas de los materiales A. M. Colliey, Antony M. Colliey, Derek J. Powney, 1977 Este texto estudia las propiedades de los materiales, utilizando el método de la teoría molecular. En muchos casos el desarrollo del tema se hará desde el punto de vista microscópico, pero concentrándose en demostrar cómo pueden explicarse los resultados obtenidos macroscópicamente.

Fundamentos de termodinámica técnica Michael J. Moran, Howard N. Shapiro, 2004 Este libro ha evolucionado a lo largo de muchos años de enseñanza de la asignatura tanto para no graduados como postgraduados. Explicaciones claras y completas, junto a numerosos ejemplos bien desarrollados, hacen el texto agradable y casi idóneo para el

Química Para El Acceso a Ciclos Formativos de Grado Superior .e-book. María Dolores Santana, 2004-05-20

Química. Fundamentos experimentales Robert W. Parry, 1973 Este libro es una versión autorizada del célebre libro original CHEM Study, cuya versión española se ha difundido por los Centros de Enseñanza donde se cursan estudios de nivel medio y preuniversitario. Tanto en este libro, como en el CHEM, la experimentación es el vehículo para presentar la Química, tal como es hoy en día. Los principios unificadores se desarrollan a partir de la observación experimental. Así, la Química aparece como una Ciencia y no como una masa de información. El objetivo principal de este libro es lograr fundamentalmente el hito de la Ciencia moderna: desarrollo del principio a partir de la observación.

Gran manual de magnitudes físicas y sus unidades. Atanasio Lleó Morilla, Lourdes Lleó Morilla, 2011-09-07 Este libro incluye, para cada una de las magnitudes físicas -La definición conceptual. -Las expresiones matemáticas principales. La unidad del Sistema Internacional, SI. -Una completa relación de todas las unidades existentes, actuales y antiguas, de la magnitud estudiada, con sus equivalencias. -Algunos valores y datos numéricos de interés. Se trata de un completísimo vademecum o handbook actualizado, referente a las magnitudes y unidades físicas que, dado el vertiginoso avance de la tecnología, venía siendo cada vez más necesario. En este momento no existe nada equivalente en la bibliografía mundial, en ningún idioma. De consulta habitual (imprescindible, podría decirse) para: -Estudiantes. -Profesores. -Ingenieros de todas las ramas. -Titulados en todas las ciencias, incluyendo Medicina. -Investigadores. -Técnicos de todos los niveles. -Economistas. -Empresarios. -Comerciantes. -Toda persona que maneje instrumentos. No parece exagerado pensar que este libro se encontrará pronto en todos los organismos y empresas de tecnología, proyectos e investigación, así como en todas las Universidades (utilizado habitualmente por profesores y alumnos), e incluso sobre la mesa de los directivos y técnicos de empresa, como manual de consulta ineludible. INDICE RESUMIDO: Breve historia del Sistema Internacional (SI). Unidades básicas del SI. Unidades derivadas. Obtención de unidades por las dimensiones. Prefijos decimales. Normas y recomendaciones. Mecánica (Partículas y sólidos). Física de fluidos y números característicos adimensionales. Fenómenos periódicos. Ondas. Acústica. Termodinámica y elasticidad. Electricidad y Electromagnetismo. Ondas electromagnéticas. Química Física, química atómica y química molecular. Física nuclear Radiaciones.

Introducción a la ciencia e ingeniería de los materiales. II William D. Callister, 1996 El principal objetivo planteado en este texto consiste en presentar los fundamentos de la Ciencia y de la Ingeniería de los materiales a un nivel comprensible para los estudiantes universitarios que han terminado los cursos introductorios de Matemáticas, Química y Física. En orden a conseguir esta se utiliza una terminología familiar para los estudiantes que se encuentran por primera vez con la Ciencia e Ingeniería de materiales y también definiendo y, posteriormente, utilizando términos no familiares.

Biología celular y molecular Harvey Lodish, 2005 Contenido I. FUNDAMENTOS QUÍMICOS Y MOLECULARES II. ORGANIZACIÓN CELULAR Y BIOQUÍMICA III. GENÉTICA Y BIOLOGÍA MOLECULAR IV. SENALIZACIÓN CELULAR V. TRÁFICO DE MEMBRANA VI. CITOESQUELETO VII. CICLO CELULAR Y CONTROL DE LA PROLIFERACIÓN CELULAR.

Física general, 2003 Un clásico entre los manuales de física universitaria, incluye todos los conocimientos que se requieren en física general. Con el objetivo de reforzar los conocimientos teóricos adquiridos en cada tema, se proponen a lo largo de todo el texto un total de 2.100 problemas cuya solución se encuentra en el libro problemas de física de los mismos autores y también publicado por Editorial Tébar.

Principios de Química. La búsqueda del conocimiento. P. W. Atkins, Loretta Jones, 2006

Química, un proyecto de la A.C.S. American Chemical Society, 2005 Este texto de Química está desarrollado y escrito por

un grupo de excelentes educadores químicos escogidos por la American Chemical Society. Poniendo un constante énfasis en el aprendizaje activo, la obra propone a los estudiantes que no lean simplemente sobre el mundo de la química, sino que lo exploren de manera personal. Química es un libro esencial para aquellos profesores que quieran incorporar a su docencia el aprendizaje basado en la investigación.

Técnicas de bioquímica y biología molecular David Freifelder, 1981-06 Este libro está dedicado a los universitarios de cursos superiores y a recién graduados con un conocimiento general de Química y Física. En algunas de las técnicas presentadas (especialmente en la sección de métodos espectroscópicos), se requiere un mayor conocimiento de Física, por lo que el estudiante puede necesitar ayuda del conocimiento del instructor sobre esta técnica, para una mayor comprensión general.

Fundamentos de la química general e inorgánica Hans Rudolf Christen, 1977 Este texto tiene como objetivo proporcionar a los estudiantes una introducción en los fundamentos teóricos de nuestra ciencia que les permita alcanzar simultáneamente una primera versión general sobre el campo de la Química inorgánica. De acuerdo con esta idea, la parte general -teoría atómica y de enlaces, termodinámica, cinética, reacciones ácido-base y redox, Química de los complejos- ocupa un espacio bastante amplio, mientras que la Química descriptiva -campo del estudio de las sustancias propiamente dicho- queda más bien en un segundo término sin formar el objetivo fundamental de la obra.

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