

# Ejercicios De Uniones Químicas Iónicas Y Covalentes

**Paul Allen Tipler**

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*Química en ejercicios* Programa UBA XXI // María José Billorou et al. Universidad de Buenos Aires, 2014-03-01 Ejercicios de química para estudiantes de UBA XXI

*Problemas y ejercicios de aplicación para química* Gabriela Mohina, Lilia Davel, 2020-09-01 Este libro brinda un compilado de problemas que responden a los temas del programa de Química que se dicta en UBA XXI. Su resolución requiere la aplicación de los conceptos fundamentales adquiridos en el estudio teórico de la materia, tales como estructura atómica y tabla periódica, uniones químicas y propiedades de las sustancias, soluciones, magnitudes atómico-moleculares y estado gaseoso, reacciones químicas, equilibrio químico, equilibrio ácido-base, soluciones reguladoras (buffer) y química orgánica. Las autoras crearon un abanico de situaciones problemáticas y de ejercitaciones con el propósito de favorecer los procesos de construcción en esta disciplina

*Química en apuntes* Gabriela Mohina, Patricia Moreno, María Gabriela Muñoz, 2020-03-01 Ligado directamente a la asignatura Química que se dicta en el Programa de Educación a Distancia de la Universidad de Buenos Aires (UBAXXI), este material se propone ofrecer a los estudiantes herramientas conceptuales que los auxilien en el recorrido por los conceptos

fundamentales asociados a esta materia.

Química 1. Fundamentos,

Curso de química general Francisco Santamaría Herrero, 1966

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Guía bibliográfica para el ingreso a las universidades, 1977

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**Conceptos y modelos para comprender la química** Darío Estrin, Sara Aldabe Bilmes, Luis Ignacio

Domenianni, 2024-03-01 Podemos definir a la química como una ciencia natural, es decir que emplea metodologías científicas para estudiar algunos fenómenos o propiedades naturales, a los que llamamos fenómenos o propiedades químicas. Los fenómenos y propiedades químicas están determinados por las propiedades de la materia estudiada, pero además tratan de conciliar la visión atómico-molecular del fenómeno con el comportamiento macroscópico que resulta de la misma sustancia. Este libro plantea un recorrido por los conceptos básicos para descubrir conjuntamente cómo pensamos y razonamos desde los átomos, las moléculas y sus interacciones con el mundo que nos rodea. Condensa, con una fuerte carga de trabajo experimental, las bases conceptuales de la Química, con especial énfasis en la comprensión de casos sencillos para describir sistemas complejos. La vida, los minerales o las estrellas, el medio ambiente, requieren miradas disciplinarias diversas, una de las cuales es la molecular y termodinámica. Con estas bases también se encaran los desafíos sociales y tecnológicos:

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Fundamentos de Química Ralph A. Burns, 2003

**Química, un proyecto de la A.C.S.** American Chemical Society, 2005 Este texto de Química está desarrollado y escrito por un grupo de excelentes educadores químicos escogidos por la American Chemical Society. Poniendo un constante énfasis en el aprendizaje activo, la obra propone a los estudiantes que no lean simplemente sobre el mundo de la química, sino que lo exploren de manera personal. Química es un libro esencial para aquellos profesores que quieran incorporar a su docencia el aprendizaje basado en la investigación.

**Biofísica y fisiología celular** Ramón Latorre, 1996

*Libros en venta en Hispanoamérica y España*, 1988

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**El Enlace Covalente** Julio Francisco Mata Segreda, 2004

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web describe some ways that time was wasted during the project that you are thinking of and how the project s schedule was adjusted or could have been adjusted to account for the wasted time your journal entry must be at least 200 words

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web identify the core agile project management principles p2p used in managing this project how would the project have been handled differently if traditional project management principles had been utilized instead

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