

Do Sophie And Keefe Kiss

M Lipman

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web the molar volume will be the volume occupied by one molar mass 32 g of the liquid expressing the density in liters instead of ml we have $v = \frac{m}{d}$ $\frac{32 \text{ g}}{1.790 \text{ g/l}} = 17.88 \text{ l}$ $\frac{10.0405 \text{ l}}{1 \text{ mol}} = 17.88 \text{ l/mol}$

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at the same temperature will have the same kinetic energy based on this principle it is found
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