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SA Adler

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can accept a pair of electrons from a base. Acids are proton donors that yield hydronium ions in water solution, or electron-pair ... *Acid: Definition and Examples in Chemistry - ThoughtCo* Jan 13, 2020 · An acid is a chemical species that donates protons or hydrogen ions and/or accepts electrons. Most acids contain a hydrogen atom bonded that can release (dissociate) to yield a cation and an anion in water. The higher the concentration of hydrogen ions produced by an acid, the higher its acidity and the lower the pH of the solution.

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