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techniques used in this lab are
quantitative transfer and vacuum
filtration with the reaction of 8 001
grams of copper ii sulfate cuso4 and 2
0153 grams of iron powder fe the goal
of this experiment was to determine the
product of copper ii sulfate with iron
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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4 stoichiometry lab the reaction of iron with copper ii sulfate - Oct 05 2022

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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period 5 in this experiment i observed
the reaction of iron with copper ii
sulfate i obtained the results to this
experiment through adding heated
cuso4 solution to the iron powder
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