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common polyatomic ions only one of them the ammonium ion is a cation the rest are anions most of them also contain oxygen atoms so sometimes they are referred to as oxyanions

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polyatomic practice 1 name or write the formula for the following polyatomic ions sulfate nitrite perphosphate hypoiodite chlorite phosphite percarbonate bromate hyposulfite permanganate carbonite CO_3^{2-} MnO_3^{1-} SO_5^{2-} BrO_2^{1-} CO_4^{2-} PO_5^{3-} ClO_1 IO_2^{1-} PO_4^{3-} NO_2^{1-} SO_4^{2-}

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formulas names and charges that cannot be modified in any way table 3 4

1 3 4 1 lists the ion names and ion formulas of the most common polyatomic ions for example NO_3^- is the nitrate ion it has one nitrogen atom and three oxygen atoms and an overall 1 charge

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion NO_3^- contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit

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2 elements in the same family make similar ions

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have characteristic formulas names and charges that should be memorized

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