

The Chilbury Ladies Choir

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 $2\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{LiCl}$ double displacement
 $3\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$ combustion
 $4\text{Pb} + \text{Fe}_3\text{O}_4 \rightarrow 4\text{PbO} + \text{Fe}$ single displacement
 $5\text{CaCO}_3 \rightarrow \text{CaO} + 5\text{CO}_2$ decomposition
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2pbs042pbs03 1o2 type of reaction
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web a chemical reaction is a process in
which one or more substances are
converted to one or more different
substances the starting substances are
called the reactants and the new
substances that form are called the
products
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give an example of a chemical reaction
leaves turning color fireworks food
spoiling 2 differentiate between a
physical and chemical change physical
does not change the identity of the
matter can be changed back easily
chemical new substance is created with
new physical and chemical