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Jiyuan Zhang

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$\text{F}_2 + 2 \text{NaF} + \text{Cl}_2 + 2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O} + \text{Pb} + \text{OH} + 2 \text{HCl} + 2 \text{H}_2\text{O} + 1 \text{PbCl}_2 + \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 + 6 \text{KBr} + \text{Al}_2\text{SO}_4 + 3 \text{CH}_4 + 2 \text{O}_2 + 1 \text{CO}_2 + 2 \text{H}_2\text{O} + \text{C}_3\text{H}_8 + 5 \text{O}_2 + 3 \text{CO}_2 + 4 \text{H}_2\text{O} + \text{C}_8\text{H}_{18} + 25 \text{O}_2 + 16 \text{CO}_2 + 18 \text{H}_2\text{O}$

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 balance the equations below  
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 $3 \text{ KClO}_3 \rightarrow 3 \text{ KCl} + 2 \text{ O}_2$   
 $2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2$   
 $2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$   
 $5 \text{ Pb} + 2 \text{ HCl} \rightarrow 2 \text{ H}_2 + \text{PbCl}_2$   
 $2 \text{ Al} + 3 \text{ Br}_2 \rightarrow 2 \text{ AlBr}_3$   
 $3 \text{ K}_2\text{SO}_4 + 4 \text{ H}_2\text{O} \rightarrow 3 \text{ K}_2\text{O} + 2 \text{ H}_2\text{SO}_4$   
 $4 \text{ CH}_4 + 7 \text{ O}_2 \rightarrow 2 \text{ CO}_2 + 2 \text{ H}_2\text{O}$   
 $8 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$

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 $1 \text{ O}_2 + 2 \text{ NH}_3 \rightarrow \text{H}_2\text{O} + 2 \text{ N}_2$   
 $2 \text{ O}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ H}_2\text{O}$   
 $2 \text{ O}_2 + 4 \text{ RbH} \rightarrow 2 \text{ RbOH} + 2 \text{ H}_2$   
 $5 \text{ Cl}_2 + 2 \text{ KBr} \rightarrow 2 \text{ KCl} + \text{Br}_2$   
 $6 \text{ N}_2 + 2 \text{ H}_2 + 3 \text{ N}_2\text{F}_6 \rightarrow 3 \text{ CaF}_2 + 8 \text{ CH}_4$   
 $2 \text{ H}_2\text{O} + \text{CO} \rightarrow \text{H}_2 + \text{C}_2\text{H}_5\text{OH}$

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 $\text{N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$   
 $\text{KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$   
 $\text{NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2$   
 $2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$   
 $\text{Pb} + \text{OH}^- \rightarrow 2 \text{ HCl} + 2 \text{ H}_2\text{O}$   
 $\text{PbCl}_2 + \text{AlBr}_3 \rightarrow 3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} + \text{Al}_2\text{SO}_4$   
 $3 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow \text{CO}_2 + 2 \text{ H}_2\text{O}$

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 $4 \text{ Fe} + 3 \text{ O}_2 \rightarrow 2 \text{ Fe}_2\text{O}_3$   
 $2 \text{ FeBr}_3 + 3 \text{ H}_2\text{SO}_4 \rightarrow 1 \text{ Fe}_2(\text{SO}_4)_3 + 3 \text{ H}_2\text{O} + 2 \text{ C}_2\text{H}_4\text{O}_2$

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 $38 \text{ Fe} + 2 \text{ O}_2 \rightarrow 3 \text{ Fe}_2\text{O}_3$   
 $2 \text{ Fe} + 3 \text{ CO} \rightarrow 2 \text{ Fe}_3\text{C} + 3 \text{ CO}_2$   
 $39 \text{ H}_2 + 3 \text{ PO}_4 \rightarrow 3 \text{ BaO} + 3 \text{ PO}_4$   
 $2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$   
 $40 \text{ Mg} + \text{N}_2 \rightarrow 2 \text{ Mg}_3\text{N}_2$   
 $4 \text{ PO}_4 + 2 \text{ Mg} \rightarrow 2 \text{ O}_7$   
 $2 \text{ NH}_3 + \text{H}_2 + \text{O}_2 \rightarrow 2 \text{ Al}_2\text{O}_3 + 3 \text{ H}_2\text{O}$   
 $41 \text{ Al} + 3 \text{ H}_2\text{SO}_4 \rightarrow 2 \text{ Al}_2\text{O}_3 + 3 \text{ H}_2\text{O}$   
 $42 \text{ P}_4 + 10 \text{ HCl} \rightarrow 3 \text{ H}_3\text{PO}_4 + 3 \text{ H}_2\text{O}$   
 $43 \text{ N}_2 + 2 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O} + 4 \text{ HNO}_2$

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