

Avete In Voi Li Fior E La Verdura

Sebastian Brünink

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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 are combined the acid is neutralized by
 the base the products of this reaction
 are an ionic compound which is labeled
 as a salt and water

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the reaction of an acid and a base is
 called a neutralization reaction
 although acids and bases have their

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 base cancel each other s chemistry to
 produce a rather innocuous substance
 water in fact the general reaction
 between an acid and a base is ce acid
 base water salt nonumber

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