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elements from lanthanum to lutetium all but one of which are
f block elements the valence electrons of these elements lie
in the 4f orbital lanthanum however is a d block element with
an

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series of chemical elements a comprises the 14 metallic
chemical elements with atomic numbers 57 70 from
lanthanum through ytterbium 2 3 4 lutetium is also
sometimes considered a lanthanide despite being a d block
element and a transition metal

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ionic character of the bonding between lanthanide cations
and other atoms as a result of this ionicity they can be
classified as hard strongly acidic cations this hard acid
nature is evident whether we consider the bonds between
lanthanide cations with oxygen nitrogen or halogen

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generally considered to be elements with atomic numbers 58
71 lanthanum to lutetium the lanthanide series is the group
of elements in which the 4f sublevel is being filled all of
these elements are metals specifically transition metals they

share several common properties

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when exposed to air react with oxygen to form an oxide
coating that tarnishes the surface for this reason these
metals are stored under mineral oil the remainder of the
lanthanides are not as reactive and some gd lu retain their
silvery metallic appearance for a long time

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abundant in the earth s crust despite their historic
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atoms can be understood in isolation indeed given the
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web to separate the lanthanides from other elements occurring with them they are chemically combined with specific substances to form lanthanide compounds with low solubility oxalates and fluorides for example a process known as ion exchange is then used to separate the lanthanides from each other in this process a solution of the

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web jan 1 1997 the lanthanides comprise the largest naturally occurring group in the periodic table there are more than 100 minerals known to contain lanthanides but the only two of commercial importance are monazite and bastnaesite the chapter discusses the production of lanthanide elements by electrolysis of fused salts and by metallothermic reduction

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