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Wolfgang Guggemos

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not to use a symbol that is already in use such as cd or cn for

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such as 98 use the average atomic mass that you calculated

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red 0 249 green 0 358 blue 0 403 2 32 g 2 35 g 2 359 2 219 2

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isotopes are atoms of the same element with the same

atomic number number of protons but different masses due

to different numbers of neutrons in the nuclei the atomic

mass of an element is the weighted average of the masses of the different isotopes

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weighted average

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