

Love And Other Demons

LM Reder

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 $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu}$
 how many moles of Ce will be produced from 23.0 g of Ce assuming CuCl_2 is available in excess
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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 $\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr}$
 $5\text{SNO}_2 + 3\text{NF}_3 \rightarrow 1\text{N}_2\text{O}_3 + \text{br}$ using the following equation
 $\text{Br}_2 + \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4 + \text{br}$

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 $\text{H}_2\text{O} + \text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr}_3 + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{NaBr} + \text{SNO}_2 + 3\text{NF}_3$

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3.00 kg of SiC answer $\text{Ce} + \text{SiO}_2 \rightarrow 3\text{C} + \text{SiC}$
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 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr}$
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