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Lauren Gardner

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phosphorus p in the center with 3 bonds to 3 other atoms
phosphorus is from the same column as nitrogen in the
periodic table meaning that p and n generally have the same
bonding structure note the lone pair dots without bonds on
top of p just like for n in the previous example for nh 3
chlorine cl

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have based on the number of valence electrons in each atom
when drawing the structure of an ion be sure to add subtract
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diagrams are diagrams that show the bonding between
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table and calculate the number of valence electrons using
the periodic table draw its lewis structure on paper build the
molecular model and record a photo list the number of bonds
and lone pairs surrounding the central atom and lastly
identify the structure geometry

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