

Dora When I Grow Up Book

B Lingard

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conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper ph for enzymes to do their work

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metals like zinc magnesium aluminium or iron acids have a low ph value i e below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is $acid + base \rightarrow water + salt$

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