

Cyrano De Bergerac Quotes

M Mark

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a function of the amount of base usually
in ml added to an acidic solution or as
the amount of acid added to a base
solution the base is added to

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bases the shape of the titration curve
for a weak acid or base depends on the
pk a or pk b of the weak acid or base
being titrated a solution ph as a
function of the volume of 1 00 m ce
naoh added to 10 00 ml of 1 00 m
solutions of weak acids with the
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word titulus which means inscription or
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base of the acid the four parts of the
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solution ph is due to the acid ionization
of hcl because this is a strong acid the
ionization is complete and the
hydronium ion molarity is 0 100 m the
ph of the solution is then $\text{ph} = \log 0.100 = 1.000$
12 50 ml

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analytical procedure used to determine

the accurate concentration of a sample
by reacting it with a standard solution
one type of titration uses a
neutralization reaction in which an acid
and a base react to produce a salt and
water

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base titration moles of base moles of
acid and the solution only contains salt
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shape of the titration curve for a weak
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the volume of 1.00 M NaOH added to 10.00 mL of 1.00 M solutions of weak acids with the indicated pK_a values

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