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web calorimeter the initial temperature of the water was 22
5 ec and the temperature of the solution after mixing was 20
4 ec on the basis of this experiment what is the heat of
solution per mole of KNO_3 f w 101 11 u the specific heat of
water is 4 184 j g k g soln 2 58 g 98 57 g 101 15 g Δt cal 20 4
ec 22 5 ec 2 1 ec

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web in a calorimeter the white solid baso 4 forms and the
temperature of the mixture increases to 28 1 c assuming that
the calorimeter absorbs only a negligible quantity of heat
and that the specific heat capacity of the solution is 4 18 j c g
and that the density of the final solution is 1 0 g ml calculate
the enthalpy change per mole

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web problem 8 5 1 6 8 5 1 6 when 50.0 g of 0.200 M NaCl aq at 24.1 °C is added to 100.0 g of 0.100 M AgNO₃ aq at 24.1 °C in a calorimeter the temperature increases to 25.2 °C as AgCl s

forms assuming the specific heat of the solution and products is 4.20 J/g °C calculate the approximate amount of heat in joules produced answer