

# Enfermedades De La Pitaya

Minjie Lin

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curve is a plot of the ph of a solution as  
a function of the amount of base usually  
in ml added to an acidic solution or as  
the amount of acid added to a base  
solution the base is added to

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bases the shape of the titration curve  
for a weak acid or base depends on the  
pk a or pk b of the weak acid or base  
being titrated a solution ph as a  
function of the volume of 1 00 m ce  
naoh added to 10 00 ml of 1 00 m

solutions of weak acids with the  
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solution ph is due to the acid ionization  
of hcl because this is a strong acid the  
ionization is complete and the  
hydronium ion molarity is 0 100 m the

ph of the solution is then  $\text{ph} = \log 0.100 = 1.000$   
0 000 ph  $\log 0.100 = 1.000$  b titrant volume  
12 50 ml

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the accurate concentration of a sample  
by reacting it with a standard solution  
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