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it turned out that aspirin had no benefit where the primary
outcome disability free survival was concerned worse it
carried a small increase in the risk of intracranial bleeding
during the five years of the trial more than a third of
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web back titration is used for 1 volatile substances e g nh_3 2 insoluble or slightly soluble substances e g caco_3 3

substances for which the quantitative reaction proceeds rapidly only in the presence of excess of reagent e g lactic acid aspirin 4 substances which decompose on heating e g

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web second you will titrate a sample of your aspirin acetylsalicylic acid with the standardized naoh to determine the moles of acid in a given weight of your product this will allow you to assess its purity assuming the aspirin is not contaminated with other acids the titration allows you to quantitatively determine the purity of your aspirin

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web pubchemdatabase aspirin cid 2244 here back titration was done to obtain the concentration of aspirin from known aspirin tablets amount of product expected was 0.5g measured yield obtained was 0.335g which shows there was an error in the determination

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