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in the values given in the problem you get  $q_{\text{water}} = 418 \text{ J/g} \times 110 \text{ g} \times 26.6 \text{ c} = 250 \text{ c} \times q_{\text{water}} = 550 \text{ J}$   $\Delta h_{\text{q water}} = 550 \text{ J}$

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